


# Henkel Pb-Free Soldering Technology

 **Technologies** No Lead

## Transitioning to a Pb-free Process

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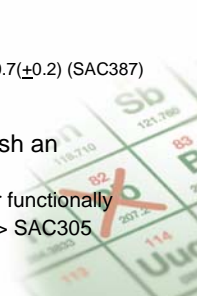


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### Marketplace - Alloy Selection

- Japan
  - still a range of materials spanning different products (result of original broad spectrum R&D?)
  - focusing down on SnAg3.0Cu0.5 (SAC305)
- Europe and Americas
  - general consensus for SnAg3.8(±0.2)Cu0.7(±0.2) (SAC387)
  - NEMI, IDEALS, IPC, SMART, NCMS
- Industry Groups trying to establish an international consensus
  - the small differences do not matter functionally
  - IPC Solder Value Council – USA -> SAC305


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### Why Pb-free?

- EU Legislation – July 2006
  - WEEE (August 13,2005) and RoHS
- Japan
  - Market Pressures
- China
  - Copy EU RoHS
  - Testing must be done by Chinese certified lab
- United States
  - State Legislation (WEEE) – California & Maine
  - CA also has a copy of the EU RoHS (Jan 2007?!?)

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### Mechanical Properties and Reliability



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## Mechanical Properties of Pb-free Alloys

- A great deal of published data on the bulk properties of Sn/Pb
  - this does not generally correlate with reliability in electronic assemblies
- Limited set of data for all the Pb-free options
  - data often contradictory need to understand how data was obtained.

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## Accelerated Testing

- Purpose of accelerated test
  - accelerate the real life failure mode
  - generate a "conversion factor" to predict lifetime
- Danger of accelerated tests
  - accelerate a failure mode that does not happen in real life
  - assume the same conversion factor applies to different situations - or different alloys

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## Factors Affecting Joint Reliability

- Assembly design - stresses & strains imposed on individual joint
  - Component/PCB size & material
  - Assembly compliance
    - LCCCs, chip components give low compliance
    - Leaded QFPs give high compliance
- Soundness of joint - interfacial defects, fillet size, standoff height, voids
- Joint microstructure
  - Solder alloy composition, dissolution from board/component finishes
  - Local cooling rate, time since joint formation

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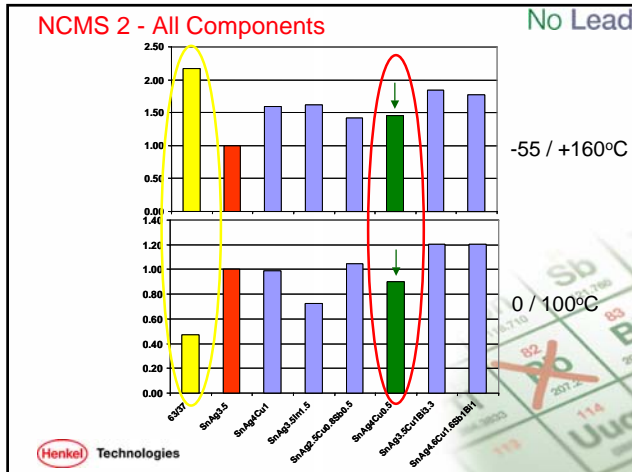
## Vibration Tests of Joints - IDEALS

Push off strengths as % of initial value	0 cycles	Random Vibration 6 hrs @20G @50-2000Hz	Thermal Shock -40/+125C	Combined RV+TS
0805 SnAgCu	100%	82%	71%	76%
1206 SnAgCu	100%	93%	92%	82%
0805 SnPbAg	100%	100%	78%	70%
1206 SnPbAg	100%	100%	65%	73%

SnAgCu slightly worse in pure vibration, better in combined thermal shock and vibration

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Assembly With Pb-free Materials

- Alloy Physical Properties
- Impact to Process
- Example of Production Study

Reliability Summary No Lead

- SnPb is better than Pb-free (SAC) under severe reliability test (-55/+160°C) conditions
- Pb-free alloys are superior to SnPb under low-cycle fatigue conditions

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Changing from Sn63 to SAC No Lead

Comparison of Alloy Properties

Alloy	Sn63	SnAg3.8Cu0.7
Melting point	183C	217C
Density (g/mm <sup>2</sup> )	8.5	7.5
Surface tension @ 260C in air	481 mNm <sup>-1</sup> *	548 mNm <sup>-1</sup> **
Wetting angle on Cu (deg)	12*	43

\* Values for Sn60Pb40  
\*\* Values for pure tin

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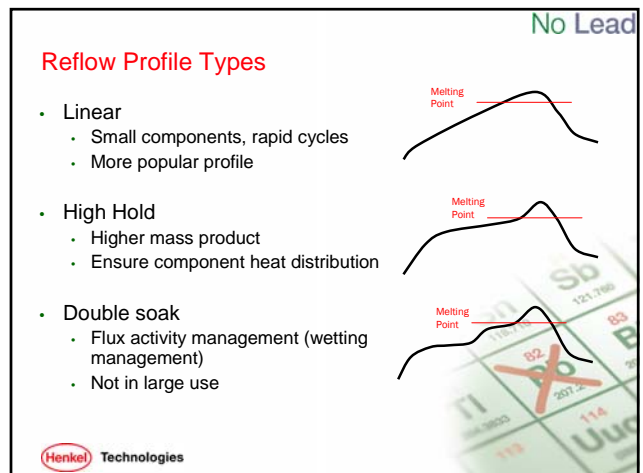
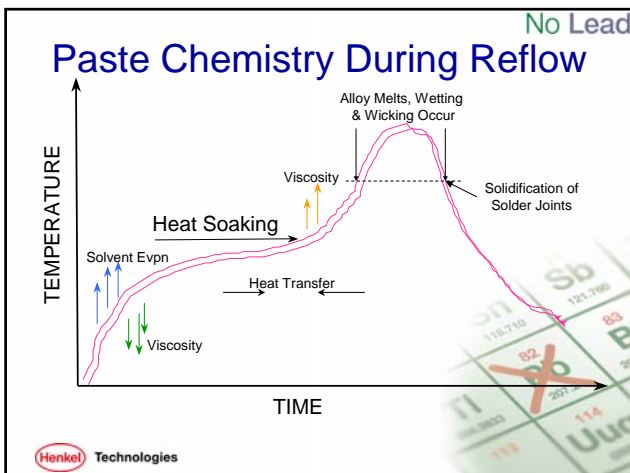
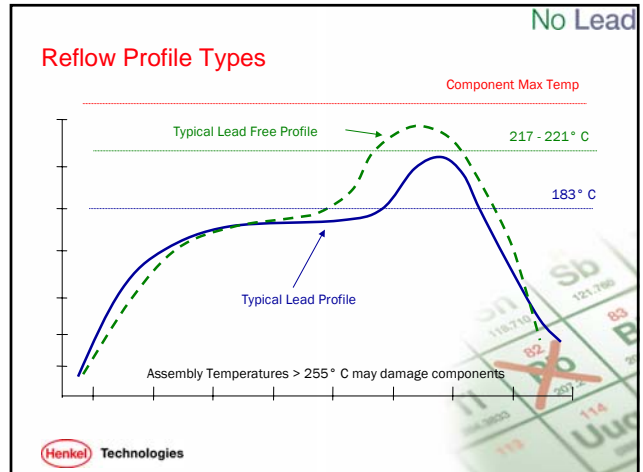
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## Physical Properties of Pb-free Alloys Affecting Assembly

- Higher melting temperature than Sn/Pb
  - Effects Reflow
  - Change for components manufacturers
  - Narrower process window
- Higher surface tension than Sn/Pb
  - Pb lowers surface tension of Sn
  - Observable effect in wetting behaviour
- Density Difference
  - Changes metal loading in paste from ~90% to ~88.5%
  - Minor impact to tack and release

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## The Pb-free Reflow Process Challenge

- Achieve good wetting on the coldest part of the circuit board
  - 235°C for 10 seconds is the accepted minimum – reports we can achieve good wetting at 229°C
- Minimize the temperature of the hottest part of the circuit board
- Keep the process stable
- Monitor the process accurately

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## Alloy Contact Angle vs. Substrate

Solder paste contact angles all in the range 20 - 25° for the same flux system

SUBSTRATE	REFLOWED ALLOY PELLETS (Sn +)					
	0.5Cu	3.5Ag	3.8Ag0.7Cu	3.5Ag0.5Sb	3.8Ag0.7Cu0.5Sb	37Pb
Cu	42	43	43	41	43	12
Ag	19	26	24	30	33	13
Sn37Pb	19	19	22	20	22	5
Sn0.7Cu	15	11	18	11	10	17
Au over Ni	9	6	10	14	5	4

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## Worst Case Scenario for Pb-Free Solder Paste

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## Wetting Angle on Copper

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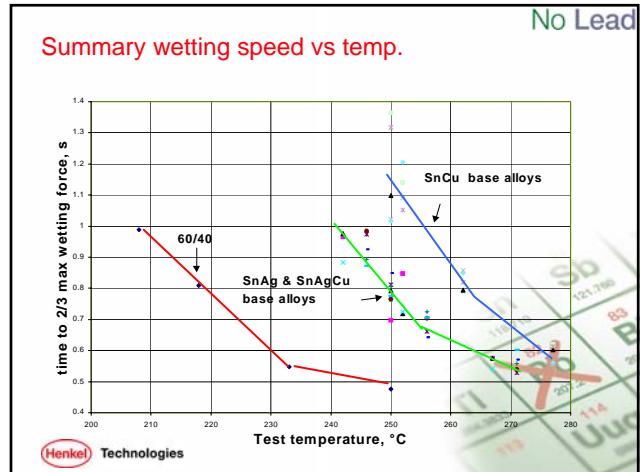
## Example of Wetting



- Solder Melting
- Wetting up the lead




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## Pb-free Alloy Wetting Behaviour

- Higher melting temperature
  - expect to need higher soldering temperatures
  - higher soldering temperature increases need for flux activity
- Wetting Balance Tests
  - quantify the **wetting rate** behaviour
- Spread Tests
  - quantify the **equilibrium wetting** behaviour




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## Can Wetting Rate be Enhanced?

- Flux activity is the main parameter
- Alloying changes
  - Sb, Bi have been proposed as wetting rate enhancers
  - Observations show the same effect of minor and impurity elements as known for Sn/Pb alloys




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## Wetting Rate and Ability

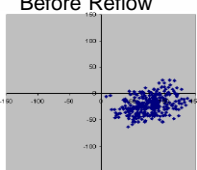
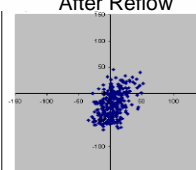
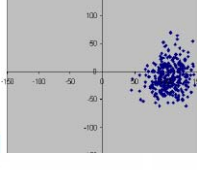
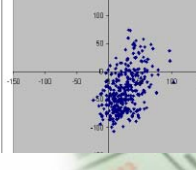
- Wetting rate effects:
  - Degree of tombstoning
  - Ability of component to self center
- Both are also related to surface tension



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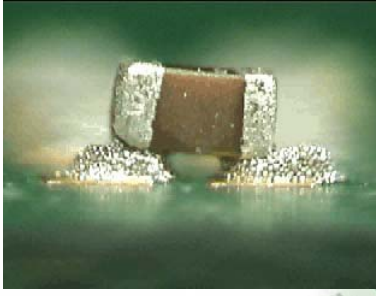
## Self-Centering Ability

	Before Reflow	After Reflow
SnPb		
Pb-free		

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## Tombstoning

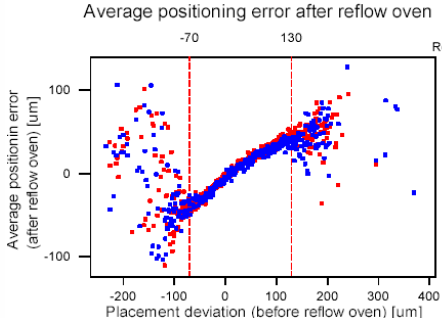


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## Component Self-Centering Process Window

Average positioning error after reflow oven



Placement deviation (before reflow oven) [um]

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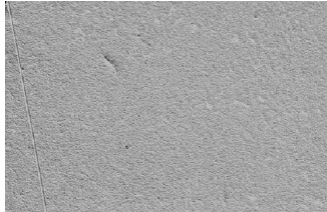
## Wetting Test Conclusions

- Wetting rates and spread for Pb-free alloys render assembly processes viable
- Surface tension of lead free alloys is higher than SnPb
  - spread is lower
- Maximum wetting force is largely independent of the alloy
- Wetting speed for SnAg and SnAgCu alloys is acceptable at lower superheat temperatures than SnCu alloys

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## Pb-free Solder Joints

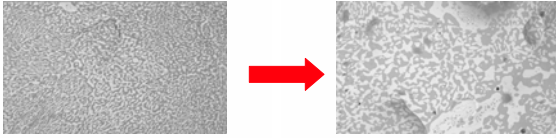


- Lead-free:
  - Sn / Ag / Cu / Au / Ni
  - More variation in surface finishes
  - OSP, IAg, ISn, ENIG
  - Needs more investigation

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## "Traditional" SnPb Solder Joints



- Currently:
  - Sn / Pb / Cu
  - Sn / Pb / Au / Ni
  - Still has lamellar structure

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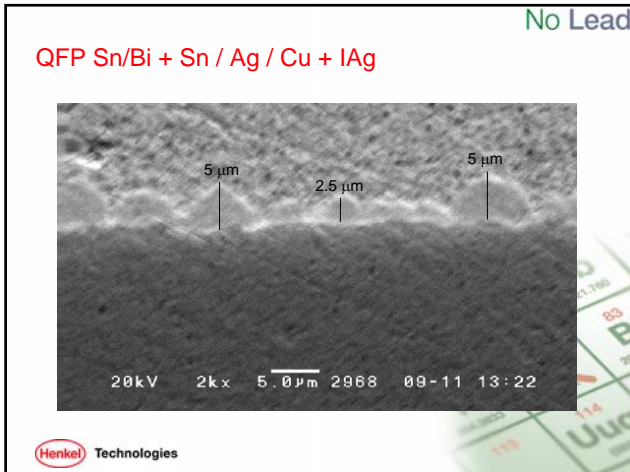
## Materials Used in This Study

- Solder Alloy
  - Sn 95.5 Ag 3.8 Cu 0.7
- Board Finishes
  - OSP
  - Immersion Tin
  - ENIG
- QFPs
  - Tin / Bismuth
  - 100% Tin
  - Tin / Copper
- Discrete Components
  - Tin/Nickel
  - 100% Tin
  - $\mu$ BGA
    - SAC 305 bumps

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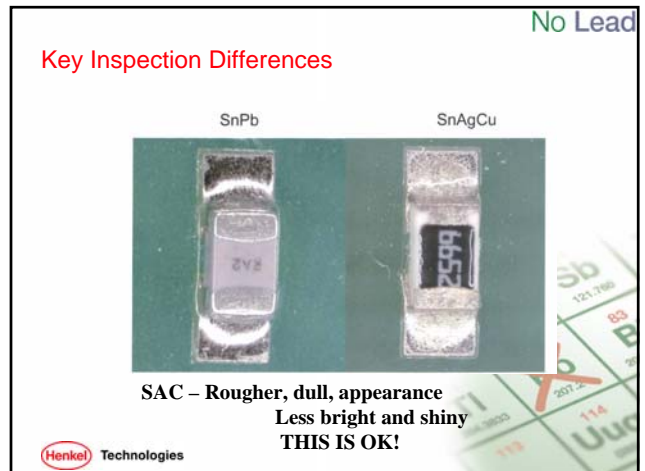
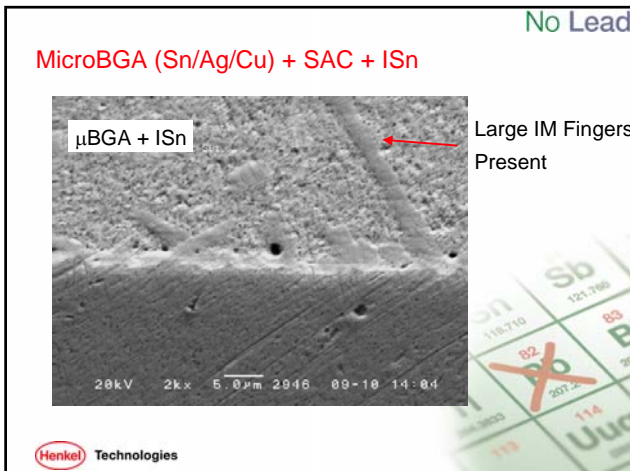
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**Beyond Engineering**  
Even if you choose the best solder paste, components and process...it is not enough

Others Involved in the Process:

- Quality
  - Need to know the differences
- Logistics
  - Transition time period / multiple lines
- Purchasing
  - Sourcing the correct materials
- Operators
  - Need to be trained to understand the differences with Pb-free


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## Key Inspection Differences

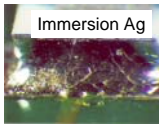


Cu OSP

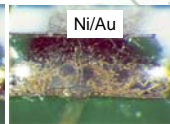


Immersion Sn

**All Pb-free**  
Different surface finishes




Immersion Ag



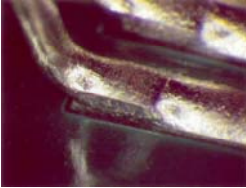
Ni/Au

Sb 121.750  
83  
114  
Uu

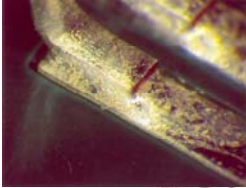


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## Wetting on cut terminations




Weaker flux reflowed in nitrogen



Stronger flux reflowed in air

**Generally poorer than Sn/Pb pastes but can be improved using flux changes and/or nitrogen**

Sb 121.750  
83  
114  
Uu



No Lead

## Solder Wetting on Pad



Sb 121.750  
83  
114  
Uu




No Lead

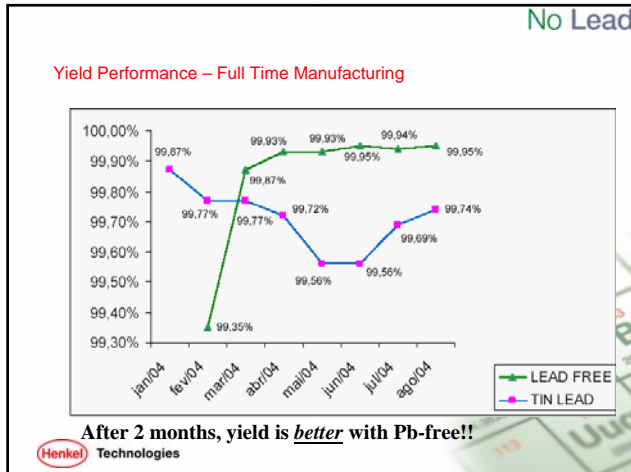
## Operator Training

- Manufacturer B working with us conducted operator training
- Review of characteristics of Pb-free Paste
  - Wetting Performance
  - Wetting Angle
  - Solder Joint Appearance
- All Operators went through 2 training courses
- 32 hours total over 2 months

Sb 121.750  
83  
114  
Uu



# Henkel Pb-Free Soldering Technology



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“Mixed Metal” Systems

- Even with exemptions full Sn/Pb may not be possible
- Mistakes will happen
- Availability of components

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Pb-free Manufacturing

- Pb-free Cell Phones have been in production for over 2 years
- Choosing the right materials is critical
- With the right training and implementation you can do as well as with Sn/Pb

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“Mixed Metal” Systems  
Pb-free “leaded” component finishes

- Pure Sn
  - Most common
  - Whiskers(?)
- Sn/Bi
  - More popular in Japan
  - Reliability Questions
- Ni/Pd/Au
  - Vendor Specific
  - Questions about wettability
- Sn/Pb + Pure Sn
  - Wetting issues
  - Reflow Profile
- Sn/Pb + Sn/Bi
  - Sn/Bi/Pb – low melt temp
- Sn/Pb + Ni/Pd/Au
  - Least affected

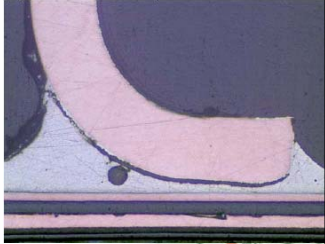
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**"Mixed Metal" Systems**  
"leaded" component finishes

- Failed SOIC lead
  - SAC 387 Paste
  - Pure-Sn Lead
  - ENIG Surface Finish
  - Components Fail after burn-in







No Lead

**"Mixed Metal" Systems**  
Pb-free bumped component finishes

- SAC Alloy
  - SAC 305
  - SAC 405
- All SACs appear to be equal
  - IPC Solder Value Council
- Many, Many, Many Questions

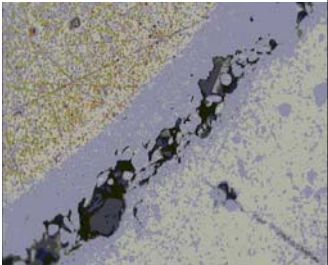





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**"Mixed Metal" Systems**  
"leaded" component finishes

- Failed SOIC lead
  - Lead 80/20 Sn/Pb
  - Failure at component lead
  - Quat eutectic (Sn/Pb/Cu/Ag) melts at 172
  - Failed in high-temp burn-in

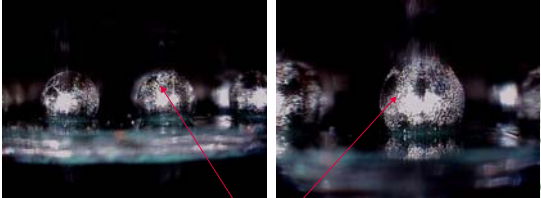





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**"Mixed Metal" Systems**  
Pb-free bumped component finishes

- Outer Appearance – BGA bumps



New BGA

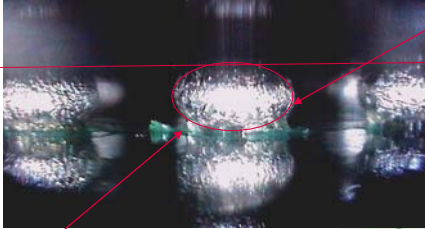


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**"Mixed Metal" Systems**  
Pb-free bumped component finishes

- Outer Appearance – BGA bumps



Ball did not melt


Lead paste Reflowed @ 210 C



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**"Mixed Metal" Systems**  
Pb-free bumped component finishes


- Pb-free BGA bumps / Sn/Pb Paste
- Metals in the solder joint:
  - Sn (bulk)
  - Ag
  - Cu
  - Pb
  - Ni
  - Au
- Not evenly distributed



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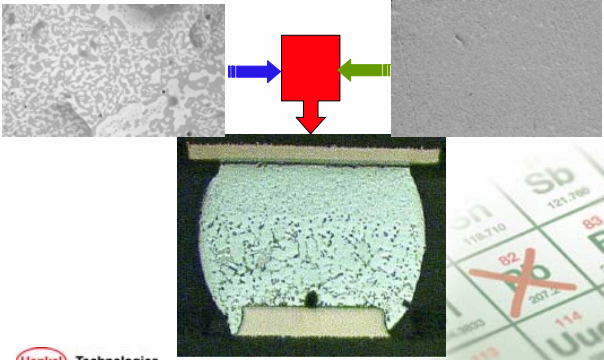

**"Mixed Metal" Systems**  
Pb-free bumped component finishes

- Pb-free BGA bumps / Sn/Pb Paste & Profile
  - Bump doesn't melt/collapse
  - Component at a higher standoff
  - Component tilt
  - Impact to reliability
- Pb-free BGA bumps & Profile / Sn/Pb Paste
  - Flux exhaustion of the Sn/Pb Paste
  - Increased voiding/blowholes
  - Poorer wetting

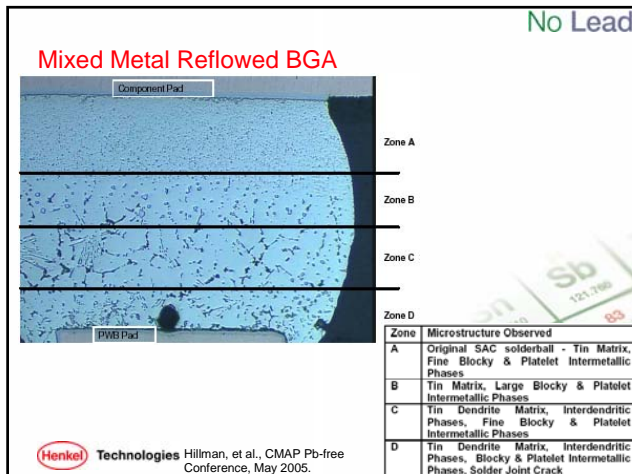


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**"Mixed Metal" Systems**  
Pb-free bumped component finishes

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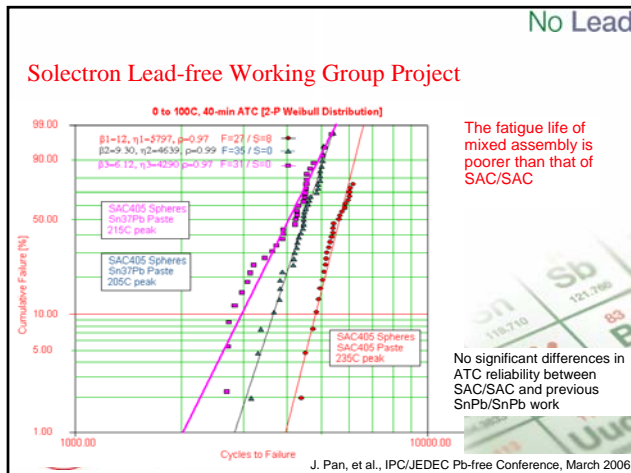
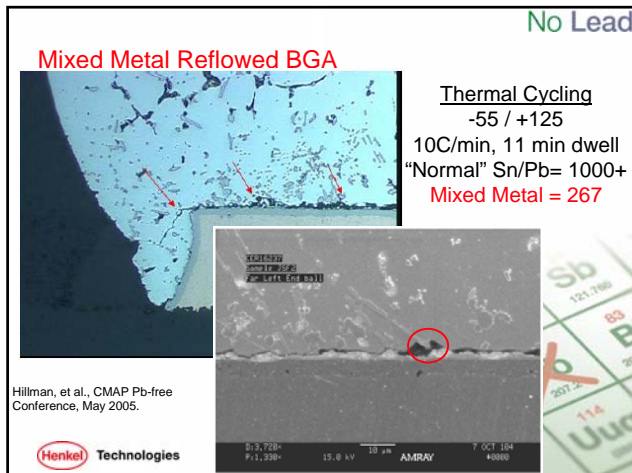
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### Final Joint Compositions and Liquidus Temp. with Sn3Ag0.5Cu Ball and Sn37Pb Paste

Pitch (mm)	Aperture Size (mil)	Stencil Thickness (mil)	Transfer Ratio (%)	Ball Dia. (mil)	% of Pb	% of Ag	% of Cu	Liquidus temp. (C)
1.27	21 (Ro)	6	100	28	3.4	2.7	0.5	216
		5	100	28	2.9	2.8	0.5	216
1.0	18 (Sq)	5	90	22	4.9	2.6	0.4	215
0.8	16 (Sq)	5	85	14	11.1	2.2	0.4	209
0.65	14 (Sq)	5	80	10	17.0	1.6	0.3	203 (200*)
0.5	11 (Sq)	4	90	10	11.9	2.1	0.3	208

\* Calculated by Dr. Ursula R. Kattner of NIST from NIST thermodynamic database

Henkel Technologies J. Pan, et al., IPC/JEDEC Pb-free Conference, March 2006



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### Effect of Pb Content

Study showed that the amount of Pb-rich phase increases and becomes coarser as the Pb impurity increases. The Pb-rich phase is the weakest region in the bulk solder, and the crack may propagate along the Pb-rich phase interface during reliability testing.

Ref. Zhu, et al., "The effect of Pb contamination on the microstructure and mechanical properties of SnAg/Cu and SnSb/Cu solder joints in SMT," *Soldering and Surface Mount Technology*, Vol. 12, No. 3, 2000, pp. 19-23.

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### Environmental Health and Safety Issues

- Solder still contains metals
- Heavy metal systems are toxic to humans
- Silver is a significant aquatic toxin
- Handle as the same as Pb-containing materials
- Good laboratory practices
  - Separate waste containers
  - Gloves, hand washing after shift/before break
- No need to report EPA/TRI

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### Conclusions

- There are success with mixed metal systems
- Reliability window is more narrow
- More work needs to be done
- iNEMI committee looking into mixed metal BGA/CSP assembly
- Other companies also investigating
- Avoid it if possible, understand the risks if required
- Purchasing needs to understand differences in components
- Incoming quality inspection?

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### Lead-Free Impact on SMT – Board Finishes

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## Requirements for a Pb Free surface Finish



- Solderable surface for Pb free alloys.
- Planar surface.
- Good shelf life.
- Can withstand 3 reflows.
- Good contact for pin probing.



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## ENIG Surface Finish

### Advantages

- Established finish
- Planar surface
- Good shelf life
- Good solderability
- Compatible with 3 reflows

### Disadvantages


- Cost
- Complex process
  - Black pad
- Gold embrittlement



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## Candidate Surface Finishes

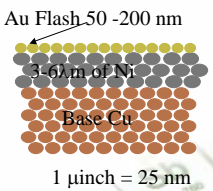
- Electroless Nickel Immersion Gold (ENIG)
- Immersion Tin (IT)
- Immersion Silver (IAG)
- OSP Copper
- Pb Free HASL (Sn/Cu/Ag)
- Electroless Nickel Electroless Paladium, Immersion Gold (ENEPIG )



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## Gold Embrittlement

Au Thickness	Stencil Thickness		
	4 Mil	5 Mil	6 Mil
50 nm	0.19%	0.16%	0.13%
100 nm	0.38%	0.32%	0.26%
200 nm	0.76%	0.63%	0.52%
500 nm	1.89%	1.58%	1.31%



Au Flash 50 -200 nm  
3-6 μm of Ni  
Base Cu  
1 μinch = 25 nm

Assumes 88% SAC 305 Paste

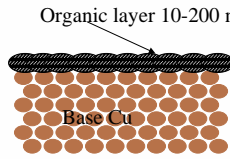




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### OSP Surface Finish



Organic layer 10-200 nm

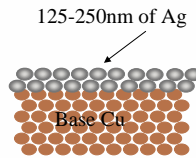
Base Cu

- Poor spread of Pb free alloys.
- Pb free OSP versions available.
- Thermally degraded
- Chemically degraded by flux.

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### Immersion Ag Surface Finish



125-250nm of Ag

Base Cu

- New surface finish (Less than 5 years)
- Ag stabilized with organic preservative
- UL electromigration concerns being addressed.

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### OSP Surface Finish

Advantages	Disadvantages
<ul style="list-style-type: none"> <li>• Low cost</li> <li>• Planar surface</li> <li>• Established finish</li> <li>• Simple intermetallics</li> <li>• Robust manufacturing process</li> </ul>	<ul style="list-style-type: none"> <li>• Low cost</li> <li>• Poor Pb free spread</li> <li>• Easily removed</li> <li>• Fragile</li> </ul>

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### Immersion Ag Surface Finish

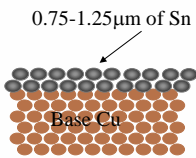
Advantages	Disadvantages
<ul style="list-style-type: none"> <li>• Planar Surface</li> <li>• Can take multiple reflows</li> <li>• Drop in replacement for HASL</li> <li>• Reasonable cost</li> <li>• Good wetting characteristics</li> </ul>	<ul style="list-style-type: none"> <li>• Limited supply base (should change)</li> <li>• Long term storage unknown</li> <li>• UL issues unresolved</li> </ul>

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## Immersion Sn Surface Finish



- Relatively thick coating of Sn.
- Compatible with Pb free alloys.
- Gained limited acceptance in Europe

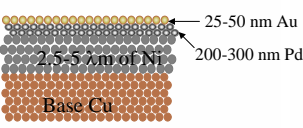
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## Other Possible Surface Finishes

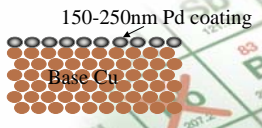
### ENEPIG

- Most expensive complex process
- Very stable highly solderable surface (no black Pad)



### Electroless Pd

- Costly
- Good solderability and shelf life



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## Immersion Sn Surface Finish

Advantages	Disadvantages
<ul style="list-style-type: none"> <li>• Planar Surface</li> <li>• Easy to rework</li> <li>• Drop in replacement for HASL</li> <li>• Reasonable cost</li> <li>• Good wetting characteristics</li> </ul>	<ul style="list-style-type: none"> <li>• Potential tin whisker problem</li> <li>• limited shelf life</li> <li>• Still limited supply</li> </ul>

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## Comparison of finishes

- There is no one finish "fits all"
- Assemblers need to evaluate surface finish pros and cons
- Things to consider:
  - Cost
  - Planarity
  - Availability
  - Storage
  - Process

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# Henkel Pb-Free Soldering Technology

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Manufacturing with a Pb-free Process

Material Compatibility


A periodic table of elements with the element Lead (Pb) crossed out with a large red 'X'. The element is located at atomic number 82 and atomic weight 207.2. Other elements visible include Sn (118.710), Sb (121.760), and Bi (208.980).

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### Adhesion versus Temperature

- Adhesion decreases with increasing temperature
- Pb free wave requires higher pot temperatures.
- It is necessary to verify adhesion at higher temperatures is sufficient to prevent lost components


A periodic table of elements with the element Lead (Pb) crossed out with a large red 'X'. The element is located at atomic number 82 and atomic weight 207.2. Other elements visible include Sn (118.710), Sb (121.760), and Bi (208.980).

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### Beyond Solder Alloy

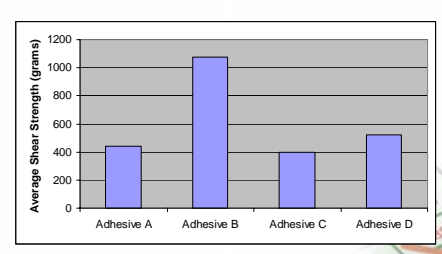
- Other materials in the process also affected by the change to Pb-free
  - Surface Mount Adhesive
  - Conformal Coating
  - Underfill
  - Substrate
  - Board Finish

A periodic table of elements with the element Lead (Pb) crossed out with a large red 'X'. The element is located at atomic number 82 and atomic weight 207.2. Other elements visible include Sn (118.710), Sb (121.760), and Bi (208.980).

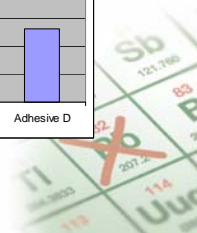
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### 0805 Shear Strength at 260C




Adhesive	Average Shear Strength (grams)
Adhesive A	450
Adhesive B	1050
Adhesive C	400
Adhesive D	550

A periodic table of elements with the element Lead (Pb) crossed out with a large red 'X'. The element is located at atomic number 82 and atomic weight 207.2. Other elements visible include Sn (118.710), Sb (121.760), and Bi (208.980).

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SMA Failure

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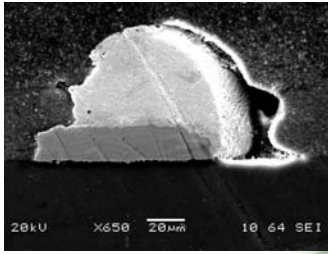
A micrograph showing a Surface Mount Antenna (SMA) component on a green PCB. The component is partially covered in a white, crystalline residue, indicating a failure in the soldering process. The background shows a portion of a periodic table with elements like Sb, Sn, and Uuq.

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Poor Underfill Adhesion

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- Example of poor underfill adhesion to flux residue



A micrograph showing a cross-section of a component where the underfill material has failed to adhere properly to the flux residue, creating a gap. Technical details at the bottom of the image include 20kV, X650, 20µm, 10 64 SEI.


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Poor Performance

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Flux Compatibility Issues

- Poor adhesion when first applied and cured
- Delamination after stress is applied




Two micrographs showing flux compatibility issues. The left image shows a component with a white, irregular residue on its surface. The right image shows a component with a vertical line of white residue, indicating delamination after stress.

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Conformal Coating Reliability

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- Need to retest with new flux systems
- SIR
- Adhesion testing
- Corrosion testing



Three images related to conformal coating reliability: a photograph of a component in a testing chamber, a micrograph of a component with a yellow conformal coating, and a photograph of a component with a yellow conformal coating.

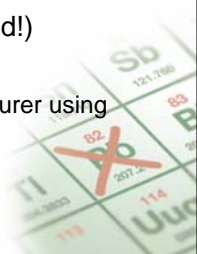
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# Henkel Pb-Free Soldering Technology

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Other Considerations

- Wave Soldering
- Rework
- Logistics
- Mixed Alloys (Transition Period!)
- Voids in Solder Joints
  - One handheld device manufacturer using underfill to meet reliability



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The slide features a background image of a periodic table with the element Lead (Pb, atomic number 82) crossed out with a large red 'X'. The text 'No Lead' is positioned in the top right corner. The Henkel Technologies logo is in the bottom left corner.

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Thank You

Questions?



The slide features a background image of a periodic table with the element Lead (Pb, atomic number 82) crossed out with a large red 'X'. The text 'No Lead' is in the top right, 'Thank You' is in the center, and 'Questions?' is in the bottom left. The Henkel Technologies logo is in the top left corner.